

C2	0.38406 (15)	-0.3931 (3)	1.26054 (12)	0.0470 (5)
C3	0.32939 (14)	-0.3408 (3)	1.18107 (10)	0.0432 (4)
C4	0.27458 (13)	-0.1270 (3)	1.16806 (10)	0.0376 (4)
C5	0.27231 (15)	0.0285 (3)	1.23787 (11)	0.0438 (4)
C6	0.32584 (15)	-0.0263 (3)	1.31783 (11)	0.0476 (5)
C7	0.22228 (13)	-0.0651 (3)	1.08119 (10)	0.0374 (4)
C8	0.16821 (13)	-0.2139 (3)	1.01730 (10)	0.0394 (4)
C9	0.13807 (13)	-0.0699 (3)	0.95103 (10)	0.0394 (4)
C10	0.08036 (15)	-0.0987 (3)	0.86229 (11)	0.0456 (5)
C11	0.03111 (15)	-0.3314 (3)	0.84962 (10)	0.0449 (5)
C12	0.4452 (2)	-0.2944 (4)	1.41627 (13)	0.0620 (6)

Table 2. Selected geometric parameters (Å, °)

O1—C9	1.349 (2)	C1—C12	1.498 (2)
O1—N1	1.408 (2)	C7—C8	1.420 (2)
N1—C7	1.307 (2)	C8—C9	1.340 (2)
N2—C11	1.140 (3)		
C9—O1—N1	108.08 (12)	C9—C8—C7	104.2 (2)
C7—N1—O1	105.54 (13)	C8—C9—O1	110.52 (14)
N1—C7—C8	111.71 (14)	N2—C11—C10	179.2 (2)
C8—C7—C4	128.72 (15)		
C3—C4—C7—C8	33.6 (2)	C8—C9—C10—C11	10.4 (3)

Refinement was on F^2 for all reflections except for three flagged by the user for potential systematic errors. The temperature of the crystal was controlled using the Oxford Cryosystem Cryostream Cooler (Cosier & Glazer, 1986). H atoms were added at calculated positions and refined using a riding model. Anisotropic displacement parameters were used for all non-H atoms, while H atoms were given isotropic displacement parameters equal to 1.2 (or 1.5 for methyl H atoms) times the equivalent isotropic displacement parameter of their attached atom.

Data collection: Siemens P3R3 system. Cell refinement: Siemens P3R3 system. Data reduction: *SHELXTL-Plus* (Sheldrick, 1990). Program(s) used to solve structure: *SHELXTL-Plus*. Program(s) used to refine structure: *SHELXL93* (Sheldrick, 1993). Molecular graphics: *SHELXTL-Plus*. Software used to prepare material for publication: *SHELXL93*.

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Lists of structure factors, anisotropic displacement parameters, H-atom coordinates, complete geometry and torsion angles have been deposited with the IUCr (Reference: BM1078). Copies may be obtained through The Managing Editor, International Union of Crystallography, 5 Abbey Square, Chester CH1 2HU, England.

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Bis[5,5-bis(4-methoxyphenyl)-3,3-dimethyl-2-tetrahydrofuranyl] Ether

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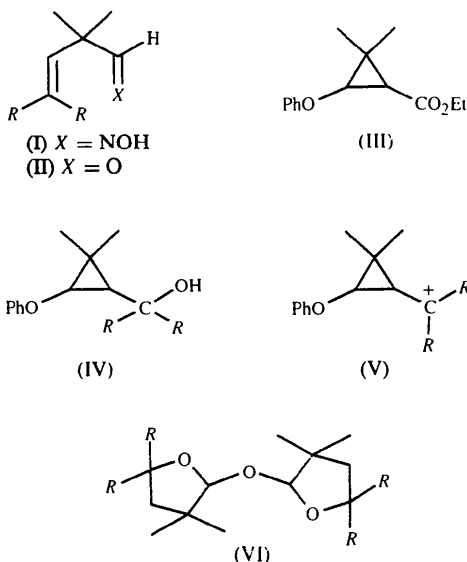
(Received 1 February 1996; accepted 25 March 1996)

Abstract

(2,2-Dimethyl-3-phenoxypropyl)bis(*p*-methoxyphenyl)methanol gave the title compound, C₄₀H₄₆O₇, an oxygen-bridged dimer, under conditions which yielded monomeric aldehydes from alcohols containing other substituents.

Comment

We are currently investigating the photochemical reactivity of a number of β,γ -unsaturated oxime derivatives, (I), under sensitized conditions using an electron-accepting sensitizer. The starting materials for this study are β,γ -unsaturated aldehydes, (II), readily made from cyclopropane esters, (III) (Julia & Baillarge, 1966; Zimmerman & Pratt, 1970). The use of a Grignard reagent converts the esters to the alcohols, (IV), which are then subjected to acid-induced ring opening. With the $R = C_6H_5$, p -^tBuC₆H₄, p -BrC₆H₄, p -ClC₆H₄ and p -CH(CH₃)₂C₆H₄ derivatives of (IV), this ring opening occurs readily. When *p*-cyanophenyl groups are the substituents [(IV); $R = p$ -CNC₆H₄], the ring opening, carried out using a mixture of acetone, concentrated hydrochloric acid and water (6:3:1) at reflux for 12 h, is an extremely slow process, presumably because conversion to the cation (V) is unfavourable. The *p*-methoxyphenyl derivative [(IV); $R = p$ -MeOC₆H₄], however, under the usual conditions (mixture of acetone, con-



centrated hydrochloric acid and water in a 4:2:1 ratio stirred at room temperature for 3 h) gave a white crystalline compound which was deposited rapidly on the sides of the flask.

It was clear from the NMR spectra that this compound was not the expected aldehyde [(II); $R = p\text{-MeOC}_6\text{H}_4$]. The identity of this product was difficult to establish from conventional techniques and even the mass spectrum provided no assistance. The crystal structure showed the product to be an oxygen-bridged dimer, (VI).

Fig. 1 shows that compound (VI) has an approximate molecular twofold axis through the O24 atom. The greatest deviation from twofold symmetry is in the methoxy groups at C11 and C41. The methoxy C15 atom is in the plane of its attached phenyl group and on the same side as C10, whereas the C45 atom is

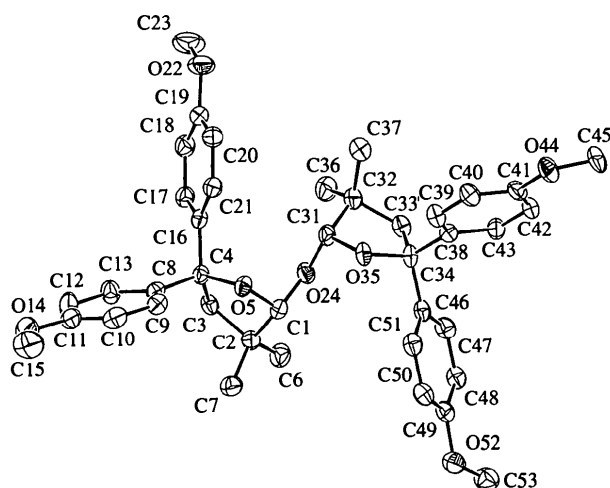


Fig. 1. A perspective view of (VI) with displacement ellipsoids plotted at the 50% probability level.

0.151 (5) Å out of its corresponding plane and faces the C42 atom. In the five-membered rings, atoms C1 and C31 are 0.571 (3) and 0.519 (3) Å, respectively, out of the planes defined by C2–C4, O5 and C32–C34, O35. There are no unusual interatomic distances or angles and no short intermolecular contacts.

Experimental

After conventional work-up and chromatography, the microcrystalline compound was dissolved in diethyl ether/hexane (1:1) at room temperature. The solvent was then allowed to evaporate slowly over a period of about a week. This yielded the product as white needles which were removed by filtration and washed with hexane. $^1\text{H NMR}$ (CDCl_3): δ 7.10–6.70 (16H, *m*, aromatic), 5.15 (2H, *s*, HCO), 3.75 (6H, *s*, OCH_3), 3.70 (6H, *s*, OCH_3), 2.50 (4H, *dd*, $J_{\text{AB}} = 10.70$ Hz, CH_2), 0.70 (6H, *s*, CH_3), 0.50 (6H, *s*, CH_3); $^{13}\text{C NMR}$ (CDCl_3): δ 157.0–126.1 (aromatic), 113.3 (OCH_3), 113.1 (OCH_3), 104.2 (OCO), 87.7 (CPh_2), 51.2 (CH_2), 44.2 [$\text{C}(\text{CH}_3)_2$], 26.0 (CH_3), 22.0 (CH_3); MS [EM, *m/z* (%): 327 (22), 311 (95), 281 (100), 267 (18), 243 (28), 227 (50).

Crystal data

$\text{C}_{40}\text{H}_{46}\text{O}_7$
 $M_r = 638.77$
Monoclinic
 $P2_1/n$
 $a = 15.2841$ (11) Å
 $b = 8.3869$ (13) Å
 $c = 28.3216$ (9) Å
 $\beta = 102.104$ (14)°
 $V = 3549.7$ (6) Å³
 $Z = 4$
 $D_x = 1.195$ Mg m⁻³
 D_m not measured

Mo $K\alpha$ radiation
 $\lambda = 0.71069$ Å
Cell parameters from 250 reflections
 $\theta = 2\text{--}25^\circ$
 $\mu = 0.081$ mm⁻¹
 $T = 150$ (2) K
Block
0.32 × 0.28 × 0.26 mm
Colourless

Data collection

Enraf–Nonius CAD-4 FAST diffractometer
Area detector scans
Absorption correction: none
11 437 measured reflections
5376 independent reflections
3159 observed reflections
[$I > 2\sigma(I)$]

$R_{\text{int}} = 0.0686$
 $\theta_{\text{max}} = 25.11^\circ$
 $h = -15 \rightarrow 16$
 $k = -6 \rightarrow 9$
 $l = -29 \rightarrow 31$
Standard reflections: see text

Refinement

Refinement on F^2
 $R(F) = 0.0449$
 $wR(F^2) = 0.1066$
 $S = 0.822$
5367 reflections
458 parameters
H atoms refined as riding model
 $w = 1/[\sigma^2(F_o^2) + (0.038P)^2]$
where $P = (F_o^2 + 2F_c^2)/3$

$(\Delta/\sigma)_{\text{max}} = 0.006$
 $\Delta\rho_{\text{max}} = 0.230$ e Å⁻³
 $\Delta\rho_{\text{min}} = -0.214$ e Å⁻³
Extinction correction: none
Atomic scattering factors from *International Tables for Crystallography* (1992), Vol. C, Tables 4.2.6.8 and 6.1.1.4)

Table 1. Fractional atomic coordinates and equivalent isotropic displacement parameters (Å²)
$$U_{eq} = (1/3)\sum_i \sum_j U_{ij} a_i^* a_j^* \mathbf{a}_i \cdot \mathbf{a}_j$$

	x	y	z	U _{eq}
C1	0.9658 (2)	0.8151 (3)	0.21391 (9)	0.0291 (6)
C2	1.0374 (2)	0.9023 (3)	0.24987 (9)	0.0287 (6)
C3	1.1166 (2)	0.7921 (2)	0.24889 (9)	0.0280 (6)
C4	1.0756 (2)	0.6231 (2)	0.24006 (8)	0.0251 (6)
O5	0.97959 (10)	0.6514 (2)	0.22509 (6)	0.0278 (4)
C6	1.0527 (2)	1.0723 (3)	0.23548 (10)	0.0398 (7)
C7	1.0120 (2)	0.8997 (3)	0.29962 (9)	0.0383 (7)
C8	1.0907 (2)	0.5195 (2)	0.28544 (9)	0.0266 (6)
C9	1.0212 (2)	0.4450 (2)	0.30093 (9)	0.0321 (6)
C10	1.0364 (2)	0.3472 (3)	0.34192 (10)	0.0381 (7)
C11	1.1220 (2)	0.3251 (3)	0.36762 (10)	0.0379 (7)
C12	1.1927 (2)	0.3993 (3)	0.35295 (10)	0.0447 (7)
C13	1.1769 (2)	0.4944 (3)	0.31226 (9)	0.0350 (7)
O14	1.14479 (15)	0.2321 (2)	0.40857 (7)	0.0593 (6)
C15	1.0736 (2)	0.1551 (3)	0.42497 (11)	0.0668 (11)
C16	1.1064 (2)	0.5327 (2)	0.19951 (9)	0.0257 (6)
C17	1.1872 (2)	0.5587 (3)	0.18668 (9)	0.0309 (6)
C18	1.2122 (2)	0.4769 (3)	0.14900 (9)	0.0349 (7)
C19	1.1554 (2)	0.3648 (3)	0.12364 (9)	0.0338 (7)
C20	1.0741 (2)	0.3335 (3)	0.13656 (9)	0.0340 (7)
C21	1.0504 (2)	0.4159 (3)	0.17368 (9)	0.0290 (6)
O22	1.17201 (13)	0.2783 (2)	0.08525 (7)	0.0498 (5)
C23	1.2537 (2)	0.3098 (4)	0.07091 (12)	0.0690 (10)
O24	0.97876 (11)	0.8500 (2)	0.16651 (6)	0.0317 (4)
C31	0.9277 (2)	0.7562 (3)	0.12933 (9)	0.0309 (6)
C32	0.9510 (2)	0.8017 (3)	0.08102 (9)	0.0337 (6)
C33	0.8913 (2)	0.9490 (3)	0.06825 (9)	0.0332 (7)
C34	0.8092 (2)	0.9209 (3)	0.09106 (9)	0.0284 (6)
O35	0.83503 (11)	0.7860 (2)	0.12377 (6)	0.0318 (4)
C36	1.0500 (2)	0.8381 (3)	0.08446 (10)	0.0469 (8)
C37	0.9223 (2)	0.6645 (3)	0.04481 (10)	0.0455 (8)
C38	0.7251 (2)	0.8709 (3)	0.05465 (9)	0.0283 (6)
C39	0.6860 (2)	0.7209 (3)	0.05677 (9)	0.0337 (7)
C40	0.6110 (2)	0.6755 (3)	0.02406 (9)	0.0348 (7)
C41	0.5715 (2)	0.7781 (3)	-0.01245 (9)	0.0315 (6)
C42	0.6080 (2)	0.9273 (3)	-0.01514 (9)	0.0338 (7)
C43	0.6843 (2)	0.9716 (3)	0.01813 (9)	0.0322 (7)
O44	0.49594 (12)	0.7214 (2)	-0.04301 (6)	0.0441 (5)
C45	0.4594 (2)	0.8158 (3)	-0.08425 (10)	0.0505 (8)
C46	0.7871 (2)	1.0605 (3)	0.12145 (9)	0.0270 (6)
C47	0.8071 (2)	1.2178 (3)	0.11305 (9)	0.0337 (7)
C48	0.7847 (2)	1.3420 (3)	0.14122 (9)	0.0325 (6)
C49	0.7409 (2)	1.3095 (3)	0.17776 (9)	0.0307 (6)
C50	0.7178 (2)	1.1530 (3)	0.18583 (9)	0.0339 (7)
C51	0.7407 (2)	1.0319 (3)	0.15792 (9)	0.0330 (7)
O52	0.71592 (12)	1.4218 (2)	0.20798 (6)	0.0393 (5)
C53	0.7177 (2)	1.5846 (3)	0.19314 (10)	0.0465 (8)

Table 2. Selected geometric parameters (Å, °)

C1—O5	1.415 (3)	O24—C31	1.411 (3)
C1—O24	1.428 (3)	C31—O35	1.415 (3)
C1—C2	1.517 (3)	C31—C32	1.532 (3)
C2—C3	1.529 (3)	C32—C33	1.534 (3)
C3—C4	1.549 (3)	C33—C34	1.545 (3)
C4—O5	1.459 (3)	C34—O35	1.463 (3)
O5—C1—O24	111.4 (2)	O24—C31—O35	111.6 (2)
O5—C1—C2	105.4 (2)	O24—C31—C32	109.4 (2)
O24—C1—C2	108.1 (2)	O35—C31—C32	105.8 (2)
C1—C2—C3	99.1 (2)	C31—C32—C33	99.9 (2)
C2—C3—C4	105.1 (2)	C32—C33—C34	105.9 (2)
O5—C4—C3	104.3 (2)	O35—C34—C33	104.3 (2)
C1—O5—C4	108.4 (2)	C31—O35—C34	109.8 (2)
C31—O24—C1	115.1 (2)		

Conventional diffraction standards were not used. The consistency was checked from successive frames of the area detector and no decay was noted.

Data collection: *MADNES* (Enraf–Nonius, 1990). Cell refinement: *MADNES*. Data reduction: *MADNES*. Program(s)

used to solve structure: *SHELXS86* (Sheldrick, 1990). Program(s) used to refine structure: *SHELXL93* (Sheldrick, 1993). Molecular graphics: *PLATON92* (Spek, 1992a) and *PLUTON92* (Spek, 1992b). Software used to prepare material for publication: *SHELXL93*.

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Lists of structure factors, anisotropic displacement parameters, H-atom coordinates, complete geometry and torsion angles have been deposited with the IUCr (Reference: LI146). Copies may be obtained through The Managing Editor, International Union of Crystallography, 5 Abbey Square, Chester CH1 2HU, England.

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2,10-Dichloro-6-(2,4-dimethylphenoxy)-dibenzo[*d,g*][1,3,6,2]dioxathiaphosphocine 6-Sulfide

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Abstract

In the crystal structure of the title compound, C₂₀H₁₅Cl₂O₃PS₂, the dioxathiaphosphocine ring adopts a boat–chair conformation, with the phosphoryl sulfide group axial and the phenoxy group equatorial. The planar chlorobenzene groups fused to the heterocyclic ring form an angle of 67.90 (10)° with one another.